## Unit \#1

| Text | Unit Objectives: |
| :---: | :---: |
| 3.3 | 1. Must be comfortable using dimensional analysis to convert <br> - between different metric units <br> - between metric units and English units |
| 3.2 3.3 | 2. Must be comfortable using table C to convert metric units when given prefixes you are not comfortable with. |
| 3.2 | 3. Must be comfortable using table D to select the correct unit of measure for different quantities. |
| 3.1 | 4. Must know what the uncertainty in measurement is. Be able to make a measurement of length, volume, mass, or temperature which is as accurate as the measuring instrument will allow. |
| 3.1 | 5. Must be able to convert from scientific notation to the written out number and back $(5,280 \mathrm{ft}=$ $5.28 \times 10^{3} \mathrm{ft}$ ). |
| 3.1 | 6. Must be able to determine the number of significant figures in a number when written out or in scientific notation. |
| 3.1 | 7. Getting answers with the correct number of significant figures after adding, subtracting, multiplying and dividing. |
| 3.4 | 8. Develop and understanding of matter and the difference between elements, compounds, and mixtures. Also know the difference between a homogenous mixture and a heterogeneous mixture. |
| 2.1 3.4 | 9. Be able to define and list properties of matter, with special attention paid to density. |
| 2.3 | 10. Know that the periodic table is a comprehensive list of all the elements and the rules for writing symbols. |
| 2.1 | 11. Know the three major states of matter and the order in which they occur. |
| 2.1 2.4 | 12. Know the difference between physical and chemical changes. |

## Essential Vocabulary

Accuracy, Atom, Binary Compound, Chemical Change, Compound, Density, Element, Gas, Heterogeneous Substance, Homogenous Substance, Liquid, Matter, Mixture, Phase, Physical Change, Significant Figures, Solid, Substance, Vapor.

## Announcements:

1. Make sure you are active in class by listening to instruction, taking good notes and solving problems at your desk.
